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### New flavonol glycosides and new xanthone from Polygala japonica

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# New flavonol glycosides and new xanthone from *Polygala japonica*

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Three new flavonol glycosides and a new xanthone were isolated from *Polygala japonica* HOUTT. with eight known compounds. Their structures were identified as 1,7-dihydroxy-3,4-dimethoxy-xanthone (1), kaempferol-7,4'-dimethyl ether (2), physcion (3), guazijinxanthone (4), rhamnetin (5), polygalin A (6), 3,5,7-trihydroxy-4'-methoxy-flavone-3-O- $\beta$ -D-galactopyranoside (7), 3,5,3'-trihydroxy-7,4'-dimethoxy-flavone-3-O- $\beta$ -D-galactopyranoside (7), 3,5,3'-trihydroxy-7,4'-dimethoxy-flavone-3-O- $\beta$ -D-galactopyranoside (9), 3,5,3',4'-tetrahydroxy-7-methoxy-flavone-3-O- $\beta$ -D-galactopyranoside (9), 3,5,3',4'-tetrahydroxy-7-methoxy-flavone-3-O- $\beta$ -D-galactopyranoside (9), 3,5,3',4'-tetrahydroxy-7-methoxy-flavone-3-O- $\beta$ -D-galactopyranoside (10), polygalin B (11), polygalin C (12). Among them, compound 4 is a new xanthone, and 6, 11 and 12 are new flavonol glycosides. Compounds 1, 4, 7 and 8 were tested for cytotoxic activity with MTT assays on five human tumor cell lines, K562, A549, PC-3M, HCT-8 and SHG-44. Compound 4 showed cytotoxic activity against all the five cell lines.

Keywords: Polygala japonica; Polygalaceae; Flavonol glycosides; Xanthone; Guazijinxanthone; Polygalin A

#### 1. Introduction

*Polygala japonica* HOUTT. (Polygalaceae), locally called 'gua zi jin', widely grows in south China. It has been used in folk medicine as expectorant, anti-inflammatory, ataractic and antibacterial agents. Many kinds of constituents have been found in this plant, such as flavones, saponins and steroids [1]. Our further investigations resulted in the isolation of three new flavonol glycosides and a new xanthone with eight known compounds; their structures were established on the basis of spectral data and chemical properties. Previous studies revealed that xanthone with terpenoid side chain such as prenyl and geranyl had significant tumor inhibitor activity [2,3]. So we evaluated the cytotoxic activity of the new xanthone as well as three known compounds, **1**, **7** and **8**, by MTT assay, a short-term *in vitro* test.

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*T-Z. Li* et al.

#### 2. Results and discussion

Compound 4 was obtained as light pink amorphous. The ion peak at m/z 424 [M]<sup>+</sup> of EI-MS is in agreement with the molecular formula  $C_{24}H_{24}O_7$ , which was confirmed by HRESI-MS, showing the  $[M + H]^+$  ion at m/z 425.1601. IR spectrum showed the presence of hydroxy  $(3509 \text{ cm}^{-1})$ , hydrogen-bonded hydroxy  $(3213 \text{ cm}^{-1})$ , hydrogen-bonded carbonyl  $(1650 \text{ cm}^{-1})$ , and benzene rings (1604, 1574 and 1472 cm<sup>-1</sup>). Its UV spectrum is similar to that of the known compound, cudraxanthone D [4]. <sup>1</sup>H-NMR spectrum showed three hydroxy protons (δ 11.94, 2H, 9.11, 1H), two aromatic protons (δ 6.47, 1H, 6.27, 1H), a methylenedioxy group ( $\delta$  6.08, 2H) and a geranyl moiety. Besides the signals of methylenedioxy and geranyl, <sup>13</sup>C-NMR spectrum demonstrated another 13 carbon signals consisted of 12 aromatic carbons and a carbonyl carbon corresponding to a xanthone moiety (see table 1) [3]. By detailed analysis of HMBC spectrum (figure 2), the two aromatic proton singlets at  $\delta$  6.47 and 6.27 were assigned to H-4 and H-7, and three hydroxyl groups at  $\delta$ 11.94, 9.11 and 11.94 were located at C-1, 6 and 8, respectively. The HMBC cross-peaks of  $H_{-1'}(\delta 3.51)$  with C-5 ( $\delta 105.6$ ), C-6 ( $\delta 162.6$ ) and C-4b ( $\delta 154.9$ ) indicated that the geranyl group was attached to C-5. The long-range correlations of protons at  $\delta$  6.08 (2H, s) with C-2 ( $\delta$  130.0) and C-3 ( $\delta$  155.5) suggested that the methylenedioxy group was fused at C-2 and C-3. Therefore, compound 4 was identified as 1,6,8-trihydroxy-2,3-methylenedioxy-5geranyl xanthone, named guazijinxanthone.

ESI-MS of compound **6**, yellow amorphous powder, showed two quasimolecular ions at m/z 499  $[M + Na]^+$  and 515  $[M + K]^+$ , which were accordant with the molecular formula  $C_{23}H_{24}O_{11}$ . Hydrolysis of **6** with 0.2 mol/l H<sub>2</sub>SO<sub>4</sub> gave aglycone and sugar. While the aglycone was identified as known compound **2** on the basis of its <sup>1</sup>H-NMR, <sup>13</sup>C-NMR spectral data [5], the sugar was identified as galactopyranose which had the same retention time as authentic sample by GC after complete acetylation. The relatively large *J* values of 8.0 Hz of the anomeric proton indicated  $\beta$ -configuration for the sugar moiety. Comparing the <sup>13</sup>C-NMR spectrum of **6** with that of **2**, glycosylation shift at C-2 (+9.5 ppm) and C-3 (-2.4 ppm) of aglycone indicated the sugar was attached to 3-OH [6]. Therefore, **6** was identified as 3,5-dihydroxy-7,4'-dimethoxy-flavone-3-*O*- $\beta$ -D-galactopyranoside, named polygalin A.

Compound **11** was obtained as yellow amorphous powder. From two quasimolecular ions at m/z 631 [M + Na]<sup>+</sup>, 647 [M + K]<sup>+</sup> in ESI-MS spectrum and <sup>13</sup>C-NMR spectra, the molecular

Table 1.	NMR data of compound 4 (CDCl <sub>3</sub> ; $\delta$ , ppm).	
δΗ	àC	

	$\delta C$	δΗ		$\delta C$	δΗ
1	143.1		9	184.5	
1-OH		11.94 (s, 1H)	9a	104.1	
2	130.0		1'	22.0	3.51 (d, J = 7.0 Hz, 2H)
3	155.5		2'	121.2	5.26 (t-like, 1H)
4	89.9	6.47 (s, 1H)	3'	139.5	
4a	153.7		4′	16.6	1.85 (d, J = 1.0 Hz, 3H)
4b	154.9		5'	40.0	2.08 (m, 2H)
5	105.6		6'	26.8	2.10 (m, 2H)
6	162.6		7′	124.0	5.04 (t-like, 1H)
6-OH		9.11 (s, 1H)	8′	132.3	
7	99.5	6.27 (s, 1H)	9′	25.9	1.65 (d, J = 1.0 Hz, 3H)
8	161.2		10′	18.0	1.58 (d, J = 1.0 Hz, 3H)
8-OH		11.94 (s, 1H)	-OCH <sub>2</sub> O-	103.0	6.08 (s, 2H)
8a	102.6		-		

New flavonol glycosides and new xanthone



Figure 1. Structure of compound 4.

formula of **11** was deduced to be  $C_{28}H_{32}O_{15}$  which had the same aglycone as **6**, namely, kaempferol-7,4'-dimethyl ether, and confirmed by comparison of <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra of **11** with those of **6**, and TLC analysis of the hydrolysate. Hydrolysis of **11** gave galactopyranose and apifuranose which were identified by the retention time of acetylated sugars by GC. The signals of two anomeric protons ( $\delta$  5.64 d, J = 7.0 Hz,  $\delta$  5.33 s) and the corresponding anomeric carbons ( $\delta$  98.9,  $\delta$  108.8) revealed the two sugar moiety were both  $\beta$  orientation [7,8]. In the HMBC spectrum, H-1<sup>III</sup> ( $\delta$ 5.33) of the apiose showed long-range correlation with C-2<sup>II</sup> ( $\delta$ 75.0) of galactose, exhibiting the apiose was attached to 2<sup>II</sup>-OH. The location of the sugar moiety was assigned to 3-OH because of the glycosylation shift of C-2 (+8.7 ppm), C-3 (-2.6 ppm) by comparing the <sup>13</sup>C-NMR spectrum with that of **2** [6]. Based upon the above evidence, the structure of **11** was characterized as 3,5-dihydroxy-7,4'-dimethoxy-flavone-3-*O*- $\beta$ -D-apiofuranosyl(1  $\rightarrow$  2)- $\beta$ -D-galactopyranoside, named polygalin B.

The ESI-MS of **12**, obtained as yellow amorphous powder, showed a quasimolecular ion at m/z 647 [M + Na]<sup>+</sup> and m/z 623 [M - H]<sup>-</sup> in negative ion mode. The mass and <sup>13</sup>C-NMR spectra suggested a molecular formula C<sub>28</sub>H<sub>32</sub>O<sub>16</sub> for **12**. Analysis of <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectrum of **12** and **11** revealed that the structures of these two compounds had the



Figure 2. Key HMBC correlations of compound 4.

403

*T-Z. Li* et al.



Figure 3. Structure of flavonoids.

same skeleton except for the substitution in the B ring, the same sugar moiety and both were 3-*O*-desmoside. By acid hydrolysis, **12** afforded 5,3'-dihydroxy-7,4'-dimethoxy-flavonol (**12a**) as aglycone [9]. So, **12** was elucidated as 3,5,3'-trihydroxy-7,4'-dimethoxyflavone-3-*O*- $\beta$ -D-apiofuranosyl(1  $\rightarrow$  2)- $\beta$ -D-galactopyranoside, named polygalin C.

The known compounds were characterized as 1,7-dihydroxy-3,4-dimethoxy xanthone 1 [10], kaem-pferol-7,4'-dimethyl ether 2 [5], physcion 3 [11], rhamnetin 5 [12],3,5,7-trihydroxy-4'-methoxy-flavone-3-O- $\beta$ -D-galactopyranoside 7 [13], 3,5,3'-trihydoxy-7,4'dimethoxy-flavone-3-O- $\beta$ -D-galac-topyranoside 8 [9], 3,5,3',4'-tetrahydroxy-7-methoxyflavone-3-O- $\beta$ -D-galactopyranoside 9 [14], 3,5,3',4'-tetrahydroxy-7-methoxy-flavone-3-O- $\beta$ -D-glucopyranoside 10 [15], by comparing with the spectral data reported in literatures.

In the MTT assay (table 4), five human tumor cell lines, including K562, A549, PC-3M, HCT-8 and SHG-44 (human glioma cell), were applied, and Topotecan was used as positive control. **4** showed cytotoxic activities against K562 with IC<sub>50</sub> value of 0.19  $\mu$ g/ml, and weak inhibitory effect against HCT-8, SHG-44, A549 and PC-3M with IC<sub>50</sub> values of 9.31, 14.41, 11.65, 24.61  $\mu$ g/ml, respectively. Compounds **1**, **7**, **8** showed no activity against all above cell lines with IC<sub>50</sub> values of more than 30  $\mu$ g/ml.

#### 3. Experimental

#### 3.1 General experimental procedures

Melting point was measured on an RY-2 melting point apparatus and is uncorrected; NMR spectra were operated on a Bruker DRX-500 spectrometer at 500 MHz for <sup>1</sup>H-NMR and 125 MHz for <sup>13</sup>C-NMR. Chemical shifts were reported with TMS as internal standard; EI-MS

#### New flavonol glycosides and new xanthone

was recorded on a Varian MAT-212 mass spectrometer and HRESI on a Q-TOF micro mass spectrometer; IR was recorded on a Bruker Vector22 spectrometer with KBr pellet; gas chromatography analysis was operated on an HP-5892 II with a FID detector, and a HP-20M (Carbowx 20M) capillary column ( $25 \text{ m} \times 0.32 \text{ mm} \times 0.3 \mu \text{m}$ ) was used; column chromatography was performed on silica gel (200-300 mesh, Yantai, China), silica gel H ( $10-40 \mu \text{m}$ , Yantai, China, macroporous resin (AB-8, Tianjin, China) and sephadex LH-20 (Pharmacia); TLC analysis was run on HSGF254 precoated silica gel plates ( $10-40 \mu \text{m}$ , Yantai, China).

#### 3.2 Plant material

Obtained as a commercial sample (purchased from Market for Traditional Material of Bozhou, Anhui province, China) and identified as *Polygala japonica* HOUTT. by Dr. Bao-kang Huang (Dept. of Pharmacognosy, College of Pharmacy, Second Military Medical University, Shanghai, China). A voucher specimen is deposited at Herbarium of School of Pharmacy, Second Military Medical University, Shanghai, China (No. 0211-1).

#### 3.3 Extraction and isolation

Dried and powdered aerial part of P. japonica (7kg) was extracted with 70% aqueous ethanol by infiltration. The solvent was evaporated under vacuum to afford 850 g crude extract. Then the extract was suspended in water and partitioned with petroleum ether, chloroform, ethyl acetate and aqua-saturated n-butol successively. Chloroform extract (65 g) was subjected to chromatography on silica gel column,  $(200 \sim 300 \text{ mesh}, 600 \text{ g})$ and eluted with petroleum ether/ethyl acetate (100:1  $\sim$  2:1) to give 10 fractions (A to J). The fraction C was separated on a silica gel column (petroleum ether/ethyl acetate 15:1, 1000 ml each fraction). The second fraction afforded compound 1 (130 mg) purified by recrystallization. The residue of the third fraction was further chromatographed on silica gel column to give compound 2 (10 mg) and compound 3 (5 mg). The fraction F was passed through a silica gel column to afford compound 4 (50 mg) (petroleum ether/ethyl acetate 10:1). Compound 5 (45 mg) was obtained from fraction J (silica gel column chromatography, petroleum ether/ethyl acetate 5:1). The ethyl acetate extraction (58 g) was separated on a silica column (200  $\sim$  300 mesh, 600 g, chloroform/methanol  $50:1 \sim 1:1$ ) to give six fractions. The third fraction afforded a yellow powder after concentration, which was purified by recrystallization in ethanol to give compound 6(180 mg). Compounds 7 (30 mg) and 8 (65 mg) were obtained from the fourth part of the ethyl acetate extraction by further chromatographed on silica gel column. Compounds 9 and 10 were obtained as mixture in a ratio of 1:1 (84 mg) from the fifth fraction by silica gel column chromatography. The n-butol extraction (210 g) was passed through a macroporous resin column (1000 g) to afford five fractions by eluting with H<sub>2</sub>O, 20%, 40%, 60% and 95% ethanol, successively. The 40% ethanol eluate of n-butol extraction was further subjected to a silica gel column chromatography (500 g) with CHCl<sub>3</sub>/CH<sub>3</sub>OH/H<sub>2</sub>O (65:25:10) as the eluent. The residues of the second and the third 1000 ml eluate were purified on a sephadex LH-20 column (100 g, CH<sub>3</sub>OH) to give compounds 11 (45 mg) and 12 (50 mg), respectively. Column separation was companied by TLC analysis on silica gel HSGF254 plates.

405

*T-Z. Li* et al.

#### 3.4 Structure and identification

**3.4.1 Guazijinxanthone** (1,6,8-trihydroxy-2,3-methylenedioxy-5-geranyl-xanthone, 4). Pink amorphous solid; mp: 178 ~ 180°C. UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ): 217 (4.75), 242 (4.58), 263 (4.69), 275 (sh 4.41), 327 (4.56), 373 (br. 4.36). IR (KBr)  $\nu_{max}$ : 3509 (-OH), 3213 (-OH), 2968, 2915, 1650 ( > C=O), 1604, 1574, 1472 (-Ph-), 1430, 1311, 1269, 1176, 1137, 1098, 1051, 993, 936, 805, 701, 575 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-NMR (see table 1). EI-MS: m/z 424([M]<sup>+</sup>), 355, 301(100), 288, 123. HRESI-MS: m/z 425.1601 [M + 1]<sup>+</sup> (calcd for C<sub>24</sub>H<sub>25</sub>O<sub>7</sub>, 425.1600).

**3.4.2 Polygalin A (3,5-dihydroxy-7,4'-dimethoxy-flavone-3-***O*-*β***-D-galactopyranoside, 6**). Yellow powder. mp: 221 ~ 223°C  $[\alpha]_D^{20}$  +33° (MeOH, *c* 0.22). UV(CH<sub>3</sub>OH)  $\lambda_{max}$  (log  $\epsilon$ ): 265 (4.14), 350 (3.99). IR(KBr)  $v_{max}$ : 3405, 2920, 1655 (>C=O), 1600, 1497 (-Ph-), 1345, 1250 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-NMR (see tables 2 and 3, respectively). ESI-MS: *m/z* 499 [M + Na]<sup>+</sup>, 515 [M + K]<sup>+</sup>. HRESI-MS: *m/z* 499.1215 [M + Na]<sup>+</sup> (calcd for C<sub>23</sub>H<sub>24</sub>O<sub>11</sub>Na, 499.1216).

**3.4.3 Polygalin B** (3,5-dihydroxy-7,4'-dimethoxyflavone-3-*O*-β-D-apiofranosyl(1  $\rightarrow$  2)β-D-galacto-pyranoside, 11). Yellow powder. mp: 146 ~ 148°C  $[\alpha]_D^{20} - 47°$  (MeOH, *c* 0.46). UV(CH<sub>3</sub>OH)  $\lambda_{max}$  (log  $\epsilon$ ): 268 (4.19), 348 (4.01). IR(KBr)  $\nu_{max}$ : 3396, 2939, 1657, 1598, 1498, 1347, 1260, 1212, 1023 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-NMR (see tables 2 and 3, respectively). ESI-MS: *m/z* 631 [M + Na]<sup>+</sup>, 647 [M + K]<sup>+</sup>. HRESI-MS: *m/z* 631.1643 [M + Na]<sup>+</sup> (calcd for C<sub>28</sub>H<sub>32</sub>O<sub>15</sub>Na, 631.1639).

**3.4.4 Polygalin C** (3,5,3'-trihydroxy-7,4'-dimethoxy-flavone-3-*O*- $\beta$ -D-apiofranosyl(1  $\rightarrow$  2)- $\beta$ -D-galac-topyranoside, 12). Yellow amorphous powder. mp: 151  $\sim$  153°C [ $\alpha$ ]<sub>D</sub><sup>20</sup> – 56° (MeOH, *c* 0.51). UV(CH<sub>3</sub>OH)  $\lambda_{max}$  (log  $\epsilon$ ): 257 (4.24), 268 (4.11, sh), 355 (3.97). IR(KBr)  $\nu_{max}$ : 3420, 2941, 1655, 1599, 1499, 1442, 1356, 1213, 1021 cm<sup>-1</sup>. For <sup>1</sup>H- and <sup>13</sup>C-NMR (see tables 2 and 3, respectively). ESI-MS: *m*/*z* 647 [M + Na]<sup>+</sup>, ESI-MS in negative mode: *m*/*z* 623 [M-H]<sup>-</sup>. HRESI-MS: *m*/*z* 647.1591 (calcd for C<sub>28</sub>H<sub>32</sub>O<sub>16</sub>Na, 647.1588).

**3.4.5** Acid hydrolysis of 6, 11, 12 in 0.05 mol/l  $H_2SO_4$ . Each compound (10 mg) was heated in 0.05 mol/l  $H_2SO_4$  (50 ml) at 90°C for 4 h. The aglycone was extracted with CHCl<sub>3</sub> and identified by <sup>1</sup>H-, <sup>13</sup>C-NMR in DMSO (see tables 2 and 3).

**3.4.6** Acetlytion of sugars. Compounds 6, 11 and 12 (5 mg each) were heated in a vial for 4 h at 80°C in 2 mol/l HCl (5 ml), respectively. After removal of solvent, the residues were dissolved in 1 ml pyridine, respectively. Then hydroxylammonium hydrochloride was added in them (1 mg each) followed by the solution were kept in a 80°C water bath for 30 min. Finally, the solution was treated with 1 ml Ac<sub>2</sub>O and kept in a 90°C water bath for another 40 min. The authentic monosaccharide samples were treated similarly as the production of hydrolysis.

	2	6	11	12	12a
Aglycone 6 8 2' 5' 6' OCH <sub>3</sub> Sugars Gal-1" 4" 6" Api-1" 4" 5"	6.36 (d, $J = 2.0 \text{ Hz}$ ) 6.75 (d, $J = 2.0 \text{ Hz}$ ) 8.18 (d, $J = 9.0 \text{ Hz}$ ) 7.12 (d, $J = 9.0 \text{ Hz}$ ) 7.12 (d, $J = 9.0 \text{ Hz}$ ) 8.18 (d, $J = 9.0 \text{ Hz}$ ) 3.87 (3H, s) 3.86 (3H, s)	$\begin{array}{l} 6.39 \ (d, \ J = 2.0  \text{Hz}) \\ 6.75 \ (d, \ J = 2.0  \text{Hz}) \\ 8.21 \ (d, \ J = 9.0  \text{Hz}) \\ 7.01 \ (d, \ J = 9.0  \text{Hz}) \\ 7.01 \ (d, \ J = 9.0  \text{Hz}) \\ 8.21 \ (d, \ J = 9.0  \text{Hz}) \\ 3.87 \ (3H, \ s) \\ 3.86 \ (3H, \ s) \\ 3.86 \ (3H, \ s) \\ 5.46 \ (d, \ J = 8.0  \text{Hz}) \\ 3.47 \ (m) \\ 3.57 \ (m) \\ 3.57 \ (m) \\ 3.40 \ 3.33 \ (m) \end{array}$	$\begin{array}{l} 6.37 \ (\mathrm{d}, J=2.0\mathrm{Hz})\\ 6.74 \ (\mathrm{d}, J=2.0\mathrm{Hz})\\ 8.25 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 7.07 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 7.07 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 8.25 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 8.25 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.86 \ (\mathrm{3H}, \mathrm{s})\\ 3.86 \ (\mathrm{3H}, \mathrm{s})\\ 3.86 \ (\mathrm{3H}, \mathrm{s})\\ 3.56 \ (\mathrm{m})\\ 3.56 \ (\mathrm{m})\\ 3.56 \ (\mathrm{m})\\ 3.41, 3.29 \ (\mathrm{m})\\ 3.38 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.38 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.36 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.36 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.36 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.50 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.50 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ 3.50 \ (\mathrm{d}, J=9.0\mathrm{Hz})\\ \end{array}$	$\begin{array}{l} 6.37 \ (\mathrm{d}, \ J=2.0\mathrm{Hz})\\ 6.71 \ (\mathrm{d}, \ J=2.0\mathrm{Hz})\\ 7.57 \ (\mathrm{d}, \ J=2.0\mathrm{Hz})\\ 7.57 \ (\mathrm{d}, \ J=2.0\mathrm{Hz})\\ 6.99 \ (\mathrm{d}, \ J=9.0,2.0\mathrm{Hz})\\ 7.91 \ (\mathrm{dd}, \ J=9.0,2.0\mathrm{Hz})\\ 3.83 \ (3\mathrm{H}, \ \mathrm{s})\\ 3.87 \ (3\mathrm{H}, \ \mathrm{s})\\ 3.87 \ (3\mathrm{H}, \ \mathrm{s})\\ 3.86 \ (\mathrm{m})\\ 3.56 \ (\mathrm{m})\\ 3.34 \ (\mathrm{m})\\ 3.43 \ 3.27 \ (\mathrm{m})\\ 3.46 \ (\mathrm{d}, \ J=6.0\mathrm{Hz})\\ 3.40 \ (\mathrm{d}, \ J=6.0\mathrm{Hz})\\ 3.40 \ (\mathrm{d}, \ J=9.0\mathrm{Hz})\\ 3.40 \ (\mathrm{d}, \ J=9.0\mathrm{Hz})\\ 3.50 \ (\mathrm{d}, \ J=9.0\mathrm{Hz})\\ 3.50 \ (\mathrm{d}, \ J=9.0\mathrm{Hz})\\ 3.50 \ (\mathrm{d}, \ J=9.0\mathrm{Hz})\\ \end{array}$	$\begin{array}{l} 6.35 \ (d, \ J = 3.0 \ Hz) \\ 6.70 \ (d, \ J = 3.0 \ Hz) \\ 7.71 \ (d, \ J = 2.0 \ Hz) \\ 7.09 \ (d, \ J = 9.0 \ Hz) \\ 7.09 \ (d, \ J = 8.0, 2.0 \ Hz) \\ 3.87 \ (3H, \ s) \\ 3.86 \ (3H, \ s) \end{array}$

Table 2.  $^{1}\text{H-NMR}$  data for compounds 2, 6, 11, 12 and 12a (DMSO, 8, ppm).

New flavonol glycosides and new xanthone

407

<i>T-Z. Li</i> et a
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Table 3. <sup>13</sup>C-NMR data for compounds 2, 6, 11, 12 and 12a (DMSO,  $\delta$ , ppm).

	2	6	11	12	12a
2	146.7	156.2	155.4	155.6	146.8
3	136.2	133.8	133.6	133.7	136.2
4	176.0	177.7	177.6	177.5	176.0
5	156.1	160.9	160.9	160.9	156.1
6	97.4	97.9	97.8	97.8	97.4
7	164.9	165.2	165.0	165.0	165.0
8	92.0	92.4	92.2	92.0	91.9
9	160.5	156.3	156.2	156.1	160.4
10	104.0	105.0	104.9	104.9	104.0
1'	123.1	122.4	122.4	122.5	123.3
2'	129.3	130.9	130.7	115.3	114.8
3'	114.0	113.7	113.6	146.0	146.2
4'	160.3	161.3	161.2	150.1	149.5
5'	114.0	113.7	113.6	111.2	111.9
6'	130.8	130.9	130.7	122.1	119.8
7-OCH3	55.9	56.1	56.0	56.0	56.0
4'-OCH3	55.3	55.4	55.3	55.6	55.7
Gal-1"		101.6	98.9	98.9	
2"		71.2	75.0	74.9	
3″		73.1	73.6	73.7	
4″		67.9	68.2	68.2	
5″		75.9	75.7	75.7	
6″		60.2	60.0	59.9	
Api-1 <sup>///</sup>			108.8	108.7	
2′′′			76.1	76.1	
3′′′			79.1	79.1	
4‴			64.2	64.3	
5′′′			73.8	73.8	

**3.4.7 Gas chromatography**. 1  $\mu$ l of the above solutions was injected into a HP-20M capillary column using N<sub>2</sub> as carrier (oven temp. 210°C) and a FID detector (detector temp. 280°C). t<sub>R</sub>: D-xylose 3.527 min, D-apiose 4.450 min, L-rhamnose 5.222 min, D-glucose 10.590 min, D-galactose 12.086 min. D-galactose was detected from compounds 6, 11 and 12, while D-apiose was detected from compounds 11 and 12.

#### 3.5 MTT assay

Compounds were tested for inhibitory activity using the microtitre MTT tetrazolium dye assay under conditions of continuous drug exposure as previously reported [16], but with some modifications. Tumor cell lines culture were diluted with fresh medium to  $4 \times 10^4$  cell/ml and plated in 96-well microplates at 100 µl/well. After 24 hours incubation at 37°C in a 5% CO<sub>2</sub> atmosphere, the test compounds of variety concentrations

Table 4.  $IC_{50}$  values ( $\mu g/mL$ ) of compounds 1, 4, 7, 8 against human tumor cell lines.

Compd	K562	SHC 11	HCT 8	1540	PC 3M
Compa.	KJ02	5110-44	IICI-0	AJ49	FC-SM
1	ND*	ND	ND	ND	ND
4	0.19	14.41	9.31	11.65	24.61
7	ND	ND	ND	ND	ND
8	ND	ND	ND	ND	ND
Topotecan	0.015	0.18	0.027	0.073	0.064

\* Not detected (IC<sub>50</sub> value > 30 µg/mL).

#### New flavonol glycosides and new xanthone

 $(10^{-2} \sim 10^2 \,\mu\text{g/ml})$  were added to the microplates in 10  $\mu$ l amounts. Then the tumor cell lines were exposed to the drugs for another 72 h. The absorbance was read on a Wellscan reader (MK-2, Labsystems, Finland) at 570 nm. Topotecan (purchased from Nanjing Tianzun Zezhong Chemical Co. Ltd., China) was used as positive reference substance with concentrations of  $10^{-3} \sim 10^2 \,\mu\text{g/ml}$ . The human tumor cell lines consisted of K562, HCT-8, A549, PC-3M, and SHG-44 (human glioma cell). The first four cell lines were purchased from ATCC. SHG-44 is preserved in Shanghai Institute for Pharmaceutical Industrial, China.

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